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Formulation and Evaluation of Glipizide Buccal Adhesive Tablets Using Natural Edible Mucoadhesives

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Abstract

In the present work,the mucodhesive tablets of Glipizide were prepared by using different concentrations of Cashewnut tree gum, Aegle marmelos gum, Moringa Oleifera as a binder. The four tablet formulation was prepared by using drug and with polymer like cashew nut tree gum, Aegle marmelos gum, Moringa Oleifers gum ratio 1:0.5,1:0.75,1:1,1:1.25 by direct compression technique. Tablets were subjected to evaluation of uniformity of weight, hardness, friability, drug content uniformity, swelling studies, surface pH study, Ex-vivo mucoadhesive time, Ex-vivo Bioadhesive strength and Invivo drug release study. Drug polymer interaction were evaluated by Fourier Transform Infrared Spectroscopy and Differential Scanning Calorimetry. All the formulations hardness, weight variation, friability and drug content values were found to be within pharmacopoeia limits. As the amount of polymer in the tablets increase, the drug release rate decreases, where as swelling index and mucoadhesive strength increases. The *in vitro* drug release of all formulations exhibits complete release of Glipizide with zero order release kinetics and followed by Higuchi mechanism. From the study it can be concluded that cashew nut tree gum, Aegle marmelos gum, Moringa Oleifera gum used as a binding agent in mucoadhesive buccal tablet.

Keywords: Glipizide, buccal tablets, Formulation, Evaluation, Extraction of natural mucoadhesive polymers.

Introduction

Among the various routes of drug delivery, the oral route is perhaps the most preferred by patients and clinicians for like1. However, peroral administration of drugs has disadvantages, such as hepatic first-pass metabolism and enzymatic degradation within the gastrointestinal tract (GIT)2. So, there has been a growing interest in the use of delivery of therapeutic agents through various transmucosal routes to provide a therapeutic amount of the drug to the proper site in body to promptly achieve and then maintain the accurate concentration, consequently other absorptive mucosa is considered as potential sites for drug administration3. Transmucosal routes of drug delivery (i.e. the mucosal linings of the oral, nasal, rectal, vaginal and ocular cavities) offer distinct advantages over peroral administration for systemic effect4. The unique environment of the oral

cavity offers its potential as a site for drug delivery, these advantages include: 1) The drug is not subjected to the destructive acidic environment of the stomach. 2) Therapeutic serum concentration of the drug can be achieved more rapidly. 3) The drug enters the general circulation without entering through the liver5. The mouth lined with a mucous membrane and among the least known of its functions is its capability of serving as a site for the absorption of drugs6. Commonly, drugs penetrate the mucous membrane by simple diffusion and are carried in the blood, which richly supplies the salivary glands and their ducts into the systemic circulation via the jugular vein7. Active transport, pinocytosis and passage through aqueous pores usually play only insignificant roles in moving drugs across the oral mucosa8. Two sites within the buccal

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cavity have been used for drug administration. Using the sublingual route, in this the medication is placed under the tongue, usually in the form of rapidly dissolving tablet9. The second anatomic site for drug administration is between the cheek and gingival, although this second application site is itself known as buccal absorption 10. The thin mucin film, which exists on the surface of the oral mucosa may provide an opportunity to retain a drug delivery system in contact with the mucosa for prolonged period, if it is designed to be mucoadhesive. Such system ensures close contact with absorbing membrane, thus optimizing the drug concentration gradient across the biological membrane and reducing the differential pathway11.In addition, it should release the drug in a unidirectional way towards the mucosa, in a controlled and predictable manner, to elicit the required therapeutic response. This unidirectional release can be achieved using bilayer device. Therefore, the oral mucosa may be a potential site for controlling or sustained drug delivery12. The permeability of the oral mucosa is low; hence the oral mucosa could be utilized to potent drugs which are required in small doses13.Bioadhesion may be defined as the state in which two materials, at least one of which is of a biological nature, are held together for extended periods of time by interfacial forces. For drug delivery purposes, the term bioadhesion implies attachment of a drug carrier system to a specific biological location14. The biological surface can be epithelial tissues or the mucous coat on the surface of a tissue15. If the adhesive attachment is to a mucous coat, the phenomenon is referred as mucoadhesion16. Generally, It has been proposed that mucoadhesion occurs in three stages. The first stage involves the formation of an intimate contact between the mucoadhesive and mucous. Secondly. mucoadhesive macromolecules swell and penetrate the mucous macromolecules, becoming physically entangled. Thirdly, these molecules interact with each other via secondary, non-covalent bonds such as hydrogen bonds 17-19.

In this study, muccoadhesive tablets of Glipizide have been developed using natural, edible mucoadhesive polymers like eagle marmelos, Cashew nut tree gum, Moringa oleifera and synthetic polymer like Ethyl cellulose each formulation had the combination. The main objective of this study is the effect of release in polymer combination and the effect of the drug: polymer ratio on drug release and other bioadhesive properties.

Materials: Glipizide was a gift sample from Arabindo Pharma Pvt. Ltd, Hyderabad, India. Aegle marmelos gum, Cashew nut tree gum and Moringa

Oliefera gum procured from Local Area. Microcrystalline cellulose and Ethyl Cellulose purchased from Qualigens fine chemicals, Mumbai. Sodium hydroxide, Sodium dihydrogen phosphate, Magnesium stearate and Talc purchased from SD fine chemicals, Mumbai. All other chemicals and reagents used were of analytical reagent grade and purchased from Himedia, Hyderabad.

Methods

Methods of preparation of Natural gums

A. Aegle marmelos gum: The fresh fruits of *Aegle* marmelos were soaked in distilled water and boiled for 5 h in a water bath until slurry was formed. The slurry was cooled and kept in refrigerator overnight. so that most of the undissolved portion was settled out. The upper clear solution was decanted off and centrifuged at 500 rpm for 20 min. The supernatant was concentrated on a water bath until the volume reduced to one third of its original volume. The solution was cooled down to the room temperature and was poured into thrice the volume of acetone by continuous stirring. The precipitate was washed repeatedly with acetone and dried at 50°C under vacuum drier. The dried gum was powdered and stored in a tightly closed container for further usage²⁰.

B. Cashew nut tree gum: The collected crude cashew nut tree gum about 100g was crushed by using mortar and pestle. The crushed gum was dissolved in water about 300ml. The solution was filtered through muslin cloth and the filtrate was collected. To the filtrate, alcohol (90% v/v) was added in 1:1 ratio and the precipitate were obtained. The precipitate was filtered and dried in a hot air oven at 45°C. 100 g of powder obtained was dissolved in 100 ml water, filtered through several folds of muslin cloth. Then the filtrate was centrifuged at 3000 rpm for 10 minutes and the supernant layer was collected, evaporated and dried to obtain solid mass. This mass was passed through sieve no. 80 and stored in an airtight container for further studies²¹.

C. Moringa oleifera tree gum: The gum was collected from incisions of trees. The gum was dried and crushed by using mortar and pestle. It is passed through sieve no.100. Dried gum was stirred in distilled water (300ml) for 4 – 5 hours at room temperature. The supernant layer was obtained by centrifugation. The residue was washed with water; this procedure was repeated for three times. Finally the supernant layer was made up to 500ml and treated with twice the volume of acetone by continuous

stirring. The precipitate material was washed with water and dried at $50 - 60^{\circ}$ C under vacuum²².

Evaluation Parameters for powdwer blend Flow properties

A. Bulk density (g/ml):About 2 gm of powder was weighed and transferred to a measuring cylinder. The bulk volume was noted. The bulk density was calculated by using following formula ²³:

Bulk Density=(Bulk Weight)/(Bulk Volume)

B. Angle of repose (θ) : The angle of repose was calculated by measuring the height and radius of the heap of powder formed as following formula:²³

$\theta = \tan^{-1}h/r$

Where, r is the radius and h is the height.

C. Carr's index (%):2 gm of powder was weighed and transferred to a measuring cylinder and it was subjected to 100 tapings. The tapped density and poured density were noted. Carr's index was calculated by the following formula²³:

Carr's Index=(Tapped Desity)-(Bulk Desity)/(Tapped Density)X100

D. Hausner's Ratio:2 gm of powder was weighed and transferred to a 25 ml measuring cylinder and subjected to 100 tapping's. The tapped density and poured density were noted. Hausner's ratio was calculated by the following formula²³:

Hausner's Ratio=(Tapped Density)/(Bulk Desity)

E.Swelling property and viscosity

Natural Mucoadhesive gum was allowed to hydrate in 25ml of distilled water at 25°C in a 25 ml graduated cylinder and volume measured at 5 minute intervals until there was no further hydration observed. The swelling property was determined at different time intervals. 1% w/v of gum solution viscosity was determined by using Broke – Field viscometer²⁴.

Preparation of Glipizide buccal tablets

Buccal tablets were prepared by direct compression procedure involving two consecutive steps. The mucoadhesive drug/polymer mixture was prepared by homogeneously mixing the drug and polymers in a glass mortar for 15 Mins. Micro crystalline cellulose, Magnesium stearate and talc were added in the blended material and mixed. The blended powder was then lightly compressed on 9 mm flat punched using sixteen station tablet compression machine (Karnavati), the upper punch was then removed and backing material ethyl cellulose was added over it and finally compressed at a constant compression force. All ingredients were dried, passed through 100 mesh sieve and mixed manually in mortar. The tablets were compressed by using sixteen station tablet machine fitted with flat faced punches and ratios of drug and all ingredients were shown in tables 1-3

Table.1.Composition of Glipizide buccal tablets formulated with different concentrations of Aegle marmelos gum

Content of tablet	$\mathbf{F_1}$ (mg)	F_2 (mg)	F ₃ (mg)	F ₄ (mg)
Glipizide	50	50	50	50
Aegle Marmelos	26	35	50	62
Microcrystalline cellulose	120	110	95	84
Magnesium stearate	2	2	2	2
Talc	2	3	3	2
Ethyl Cellulose	50	50	50	50
Total weight (mg)	250	250	250	250

Table.2. Composition of Glipizide buccal tablets formulated with different concentrations of cashew nut tree

gum									
Content of tartarate	F_5 (mg)	F_6 (mg)	$\mathbf{F_7}$ (mg)	F_8 (mg)					
Glipizide	50	50	50	50					
Cashew nut tree gum	26	40	50	60					
Microcrystalline cellulose	120	105	95	86					
Magnesium stearate	2	3	2	2					
Talc	2	2	3	2					
Ethyl Cellulose	50	50	50	50					
Total weight (mg)	250	250	250	250					

Table.3. Composition of Glipizide buccal tablets formulated with different concentrations of moringa oleifera									
gum									
	Content of tablet		$F_{10}\ (mg)$	F_{11} (mg)	$\mathbf{F}_{12}\left(\mathbf{mg}\right)$				
	Glinizida	50	50	50	50				

Content of tablet	F_9 (mg)	F_{10} (mg)	F_{11} (mg)	$F_{12}\left(mg\right)$
Glipizide	50	50	50	50
Moringa oleifera gum	26	35	50	60
Microcrystalline cellulose	120	115	96	83
Magnesium stearate	2	2	2	3
Talc	2	2	2	4
Ethyl Cellulose	50	46	50	50
Total weight (mg)	250	250	250	250

Evaluation of tablets

- **A. Hardness:** Hardness of tablet is determined by using the Monsanto hardness tester²⁶.
- **B.** Weight variation: Formulated tablets were tested for weight uniformity, 20 tablets were weighed collectively and individually. From the collective weight, average weight was calculated. The percentage of weight variation was calculated by using the following formula²⁶.

%Weight variation=(Average weight)-(Individual weight)/(Average weight)X100

C. Friability: The Roche friabilitor apparatus was used to determine the friability of the tablets. About 20 tablets were selected, dedusted and weighed. Then they were placed in a drum and rotated at 25 rpm for 4 minutes. Then tablets were dedusted to remove dust and reweighed. The percentage friability was calculated by the given formula²⁶.

% Friability=(Initial weight)-(Final weight)/(Initial weight)X100

- **D. Drug content:** Twenty tablets were collected and powdered. The powder equivalent to 50mg of the drug was weighed accurately, dissolved in 100ml of phosphate buffer pH 6.8. The solution was filtered, suitably diluted and an aliquot was analyzed at 224nm by using UV-spectrophotometer²⁷.
- **E.** *In-vitro* **dissolution test:** The release of Glipizide from the tablet was studied using USP Type II paddle apparatus. The drug release profile was carried out in 500 ml of 6.8 pH phosphate buffer maintained at $37 \pm 0.5^{\circ}$ C temperature at 50 rpm. 5 ml of sample was withdrawn at regular time intervals. The samples were analyzed at 224 nm by UV spectrophotometer²⁸.
- **F. Surface pH study:** The tablet was allowed to swell by keeping in contact with 1 ml of distilled water for

2hrs at room temperature. The pH measured was by bringing the electrode in contact with the surface of the tablet an allowing to equilibrate for 1 min²⁹.

G. Swelling study: Three buccal tablets were weighed individually (W_1) and placed separately in 2% agar gel plates at $37\pm1^{\circ}$ C. After every 2h time interval until 6h the tablet was removed from the Petri dish and excess surface water was removed carefully with blotting paper. The swollen tablet was then reweighed (W_2) and the swelling index (SI) was calculated using the formula given in equation³⁰.

Swelling index = $(W_2-W_1)/W_1 \times 100$

Where, W_1 = initial weight of the tablet, W_2 = final weight of the table

- **H. Ex-vivo mucoadhesive time:** The ex-vivo mucoadhesion time was examined after application of the buccal tablet on freshly excised goat buccal mucosa which was obtained from the slaughter house. The fresh goat buccal mucosa was tied on the glass slide and buccal tablet was pasted to the goat buccal mucosa by applying a light force with a fingertip for 30sec. The glass slide was then dipped down in the beaker, which was filled with 200ml of the phosphate buffer pH 6.8 maintained at 37±1°C. After 2min, stirring was applied by a magnetic stirrer slowly to stimulate the buccal cavity environment and tablet adhesion was maintained for 10h. The time for the tablet to detach from the goat buccal mucosa was recorded as the mucoadhesion time³¹.
- **I.** *Ex-vivo* **Bioadhesive strength:** *Ex-vivo* bioadhesive strength of the buccal tablets was measured by the modified physical balance method. The fresh goat buccal mucosa was obtained from the slaughter house was cut into pieces and washed with the phosphate buffer pH 6.8. The tablet was stick to the lower side of the second glass slide with glue. The both pans were balanced by adding an appropriate weight on the left-

hand pan. The glass slide with mucosa was placed with appropriate support, so that the tablet touches the mucosa. Previously weighed beaker was placed on the right hand pan and water equivalent to weight was added slowly to it until the tablet detach from the mucosal surface. The weight equipped to detach the tablet from the mucosal surface gave the bioadhesive strength. The experiment was performed in triplicate and the average value was calculated³².

Force of adhesion(N)=(Mucoadhesive strength)X(9.1)/(1000)

J. Infrared Spectral Analysis: Fourier Transform Infrared (FTIR) Spectroscopy studies were used for the evaluation of physicochemical compatibility and interactions, which helps in the prediction of interaction of the drug with gum, diluents and lubricants used in tablet formulations. In the present study 1:1 ratio was used for preparation of physical mixtures and analyzed for compatibility studies³³.

H. Differential Scanning Calorimetry Study: Differential Scanning Calorimetry of Glipizide and optimized formulations was recorded between 30.0°C to 300.0°C at the rate of 20.0°C per minute under the environment of nitrogen³⁴.

Results and Discussion

Preparation of Natural gums

Natural gums of plant origin have been used widely as demulcent because of their unique properties to bind to the mucus membrane. The selection of the materials

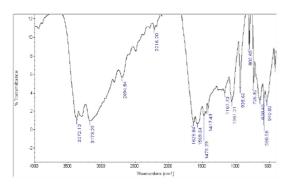


Fig.1. FTIR spectrum of Glipizide

for the current investigation was based on their edibility, blandness, availability and economics.Preparation of water-soluble components from the natural edible sources was carried out by cold/hot aqueous extraction process followed by the organic solvent precipitation. The selection of the process was based on previous literature giving utmost importance to preserve the components against thermal, enzymatic and hydrolytic degradation. The organic solvents used for precipitation can be recovered back by fractional distillation, making the process more economical. The processes used were found to be effective in the selective preparation of the interested constituents and the yielded components possessed good handling properties.

FT-IR spectrum and DSC Study

The FT-IR spectrum did not show the presence of any additional peaks for new functional groups, indicating no chemical interaction between drug and polymers. DSC thermogram showed that there was no any major difference in onset temperature and peak temperature, when compared with pure drug thermogram results are shown in figure numbers 8-9. No interaction was found between drug and polymers. From the DSC results it was observed that the characteristic peak of drug is not observed in the drug and polymer mixer. Hence it indicates the physical nature of the drug is not changed in the formulation. Therefore, results showed that there is no significant change in the chemical integrity of the drug, indicating no interaction between the drug molecule and polymers results were shown in figures 1-7.

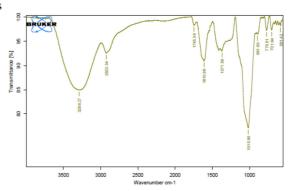


Fig.2.FTIR spectrum of Aegle marmelos gum

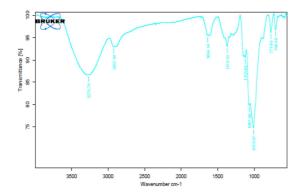
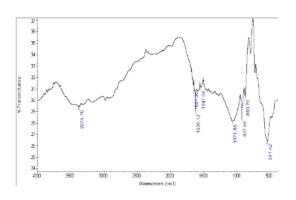
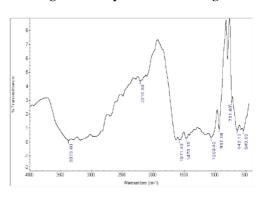


Fig.3: FTIR spectrum of Cashew nut tree gum

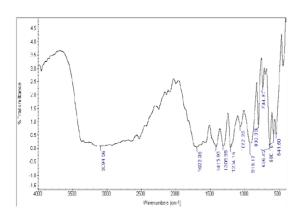
Fig.4: FTIR spectrum of Moringa oleifera gum





Fi.5:.FTIR spectrum of Glipizide buccal tablets prepared with Aegle marmelos gum

Figure 6 FTIR spectrum of Glipizide buccal tablets prepared with Cashew nut tree gum



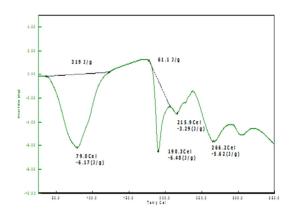


Fig. 7: FTIR spectrum of Glipizide buccal tablets prepared with Moringa oleifera gum

Fig.8: DSC thermogram of Glipizide

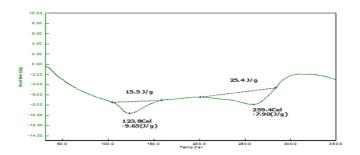


Fig.9: DSC thermogram of the Glipizide+ Polymer mixer

Evaluation Parameters

Table 4 represents the physical properties of the granules used for the preparation of tablets. The flow properties such as angle of repose, Hausner's ratio, Carr's index, Bulk density and Tapped density are considered as indirect measurements of powder flowability. Hausner's ratio is indicative of inter- particular friction; the Carr's index shows the propensity of a material to diminish in volume. As the values of these indices increase, the flow of the powder decreases. All parameter values are within the satisfactory limit compared with the standard values shown in tables 5-12.

Table. 4: Micromeritic properties of formulations blend of Glipizide buccal tablets prepared with different concentrations of Aegle Marmelos gum

	Evaluation parameters							
Formulation	Bulk density	Tapped density	Compressibility	Hausner's	Angle of			
	(g/ml)	(g/ml)	index (%)	Ratio	Repose (θ)			
F_1	0.419 ± 0.018	0.503 ± 0.20	14.16 ± 0.59	1.19 ± 0.012	28.04 ± 0.12			
F_2	0.429 ± 0.021	0.506 ± 0.025	14.83 ± 0.46	1.18 ± 0.019	28.96 ± 0.17			
F_3	0.412 ± 0.021	0.510 ± 0.031	14.23 ± 0.51	1.18 ± 0.013	28.02 ± 0.18			
F_4	0.467 ± 0.018	0.561 ± 0.021	14.66 ± 0.44	1.16 ± 0.012	28.31 ± 0.18			

Table. 5. Swelling property values of Aegle marmelos gum

Natural gum	After 5 min(ml)	After 10min(ml)	After 15 min(ml)	After 20 min(ml)	After 25 min(ml)	After 30 min(ml)	After 35 min(ml)
Aegle marmelos gum	0.8	0.9	1.2	1.4	1.5	1.5	1.5

Table. 6. Visocsity of 1% W/V dispersion of Aegle marmelos gum

S.NO	POLYMER	VISCOCITY (cps)
1	1% w/v of aegle marmelos gum	2753.15

Table.7.Micromeritic properties of formulations blend of Glipizide buccal tablets prepared with different concentrations of cashew nut tree gum

	Evaluation parameters								
Formulation	Bulk density	Tapped density	Compressibility index	Hausner's	Angle of Repose				
	(g/ml)	(g/ml)	(%)	Ratio	(θ)				
F_5	0.439 ± 0.018	0.512 ± 0.026	14.24 ± 0.71	1.16 ± 0.011	24.02 ± 0.22				
F_6	0.445 ± 0.011	0.522 ± 0.019	13.94 ± 0.52	1.17 ± 0.08	25.22 ± 0.16				
F_7	0.478 ± 0.017	0.580 ± 0.023	17.58 ± 0.45	1.21 ± 0.010	27.36 ± 0.15				
F_8	0.496 ± 0.015	0.594 ± 0.020	16.49 ± 0.56	1.19 ± 0.14	28.85 ± 0.18				

Table.8. Swelling property of Cashew nut tree gum

Natural gum	After 5min (ml)	After10min (ml)	After 15min (ml)	After 20 min (ml)	After 25 min (ml)	After 30min (ml)	After 35min (ml)
Cashew nut tree gum	0.7	0.8	1.1	1.3	1.4	1.5	1.5

Table.9: Visocsity of 1% W/V dispersion of Cashew nut tree gum

S.NO	POLYMER	VISCOCITY (cps)
1	1% w/v of cashew nut tree gum	2186.29

Table 10: Micromeritic properties of Glipizide buccal tablets formulated with different concentrations of Moringa oleifera gum

	Evaluation parameters							
Formulation	Bulk density (g/ml)	Tapped density (g/ml)	Compressibility index (%)	Hausner's Ratio	Angle of Repose (θ)			
F ₉	0.426 ± 0.016	0.502 ±0.021	15.13 ±0.57	1.17 ±0.010	23.12 ± 0.18			
F_{10}	0.452 ±0.019	0.543 ± 0.023	16.75 ± 0.53	1.20 ± 0.012	27.46 ± 0.15			
F_{11}	0.469 ± 0.021	0.571 ± 0.022	17.86 ±0.46	1.19 ± 0.013	$28.12 \pm 0.0.12$			
F_{12}	0.478 ± 0.023	0.580±0.018	17.58 ±0.49	1.21 ±0.09	29.30 ± 0.18			

Table.11: Swelling property of Moringa oleifera gum

Natural gum	After 5 min(ml)	After 10 min(ml)	After 15 min(ml)	After 20 min(ml)	After 25 min(ml)	After 30 min(ml)	After 35 min(ml)
Moringa oleifera gum	0.6	0.7	0.8	1.0	1.1	1.2	1.3

Table.12. Visocsity of 1% W/V dispersion of Moringa oleifera gum

S.NO	POLYMER	VISCOCITY (cps)
1	1% w/v of Moringa oleifera gum	1546.95

Preparation and Evaluation of Glipizide buccal tablets

Mucoadhesive buccal tablets of Glipzide with Aegle marmelos gum were prepared by using different drug: gum ratios. The results of the physical characterization of tablets are summarized in Table 13. All the formulations hardness, weight variation, friability and drug content values were found to be within pharmacopoeia limits. The swelling behavior is important for bioadhesion. Water sorption increases with an increase in the concentration of hydrophilic polymers. Swelling index, Mucoadhesive strength and *Ex-vivo* residence time were shown in Table 14.

The Aegle marmelos gum swells slowly and dissolves in the presence of water. As hydrophilicity of the hydrogel increases, the interaction between water and hydrogel will increase too; this facilitates water diffusion and leads to greater swelling. The surface pH

was determined in order to investigate the possibility of any side effects, in the oral cavity as acidic or alkaline pH was bound to cause irritation to the buccal mucosa. Surface pH of all formulations was found to be in the range of 6.32 to 6.84 which were nearer to the salivary pH 6.8. Hence it was assumed that these formulations do not cause any irritation to the mucous layer of the oral cavity. Mucoadhesion is determined by Mucoadhesive strength and duration of mucoadhesion. Formulation F₁-F₄ shows good mucoadhesive strength. As the viscosity gum increases swelling increases and mucoadhesion force depends on the swelling of the gum. This improves the consolidation step that increases the mobility of molecule and facilitates the interpretation with mucus layer, thus mucoadhesion increases. F4 shows maximum mucoadhesive strength this is due to the tremendous increase in viscosity.

Table.13. Physical properties of Glipizide buccal tablets formulated with different concentrations of Aegle Marmelos gum

1,141,111,102, 84,111					
Th 1.4°	Parameters				
Formulation	Weight variation (mg)	Hardness (kg/cm ²)	Friability (%)	Drug content (%)	
$\overline{F_1}$	250 ± 1	4.2 ± 0.01	0.22	99.13	
F_2	250 ± 1	4.3 ± 0.03	0.31	99.37	
F_3	250 ± 1	4.3 ± 0.02	0.32	99.55	
F_4	250 ± 1	4.2 ± 0.01	0.38	99.45	

Table .14. Mucoadhesion strength, swelling index, retention time, and surface pH of buccal tablets prepared with different concentrations of Aegle marmelos gum

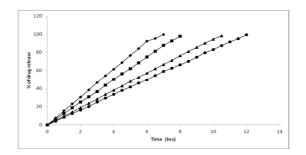
	-			
Formulation	Swelling index	Ex-vivo mucoadhesion time	Ex-vivo bioadhesive strength	Surface pH
$\overline{F_1}$	8.13 ± 3.68	4 hours 45 minutes	16.19 ± 0.28	6.33 ± 0.07
F_2	8.91 ± 3.07	6 hours 20 minutes	16.78 ± 0.31	6.45 ± 0.05
F_3	10.17 ± 7.62	8 hours 15 minutes	17.12 ± 1.25	6.44 ± 0.08
F_4	11.71 ± 6.85	10 hours 50 minutes	18.18 ± 1.36	6.72 ± 0.06

The ex-vivo residence time was determined using USP disintegration apparatus. Among the four formulations subjected for this study F_4 showed maximum residence time of 10.5 Hrs. It was found that an increase in concentration of polymer increases the residence time. This was mainly due to the strong mucoadhesion nature which of the polymer used. The results of *in vitro* drug release studies of different formulation were shown in table 15 and Figure 10. Tablet formulations prepared by using drug and gum in ratios of 1:0.5, 1:0.75 1:1, and 1:1.25 shown drug release for a period of 7 hours, 8 hours, 10.5 hours and 12 hours respectively. The initial burst release decrease with

increase in concentration of gum. To ascertain the mechanism of drug release, the dissolution data were analyzed by zero order, first order, Higuchi and Peppas equations. The correlation coefficient values (r) and dissolution kinetics values were shown in table 16. Amount of drug release versus time curves exhibited straight line for the formulations and confirmed that the release rate followed zero order release kinetics as shown in figure 11 percentages of drug release versus the square root of time curves shows linearity and proves that all the formulations followed Higuchi mechanism as shown in figure 12.

Table 15. In vitro release data of Glipizide buccal tablets prepared with different concentrations of Aegle

	Marmelos gum					
Time	F ₁ (%Drug	F ₂ (%Drug	F ₃ (%Drug	F ₄ (%Drug		
(hrs)	Release)	Release)	Release)	Release)		
0	0	0	0	0		
0.5	7.55 ± 0.05	06.10±0.07	4.75±0.10	04.03±0.09		
1	15.45 ± 0.07	12.45±0.09	9.35±0.07	08.25±0.06		
1.5	23 ± 0.10	18.65±0.17	14.10±0.09	12.43±0.10		
2	30.25±0.09	24.91±0.14	19.00±0.14	16.19±0.13		
2.5	38.15±0.06	30.65±0.07	23.46±0.16	20.05±0.15		
3	46.35±0.08	36.25±0.09	28.25±0.12	24.91±0.10		
3.5	53.85±0.11	43.45±0.05	33.3±0.11	29.25±0.07		
4	61.25±0.14	49.85±0.08	37.95±0.15	33.21±0.11		
4.5	68.75±0.10	56.05±0.11	42.35±0.13	37.65±0.16		
5	76.55±0.13	61.85±0.15	47.65±0.09	41.45±0.12		
5.5	84.15±0.16	68.43±0.12	52.25±0.11	45.73±0.14		
6	92.25±0.12	74.83±0.05	57.08±0.14	49.35±0.09		
6.5	95.45±0.08	81.03±0.07	61.75±0.05	54.17±0.15		
7	99.75±0.15	87.75±0.10	66.45±0.10	58.80±0.08		
7.5	-	92.71±0.14	71.21±0.16	62.32±0.13		
8	-	97.7±0.16	76.22±0.13	66.11±0.11		
8.5	-	-	80.84±0.08	70.15±0.07		
9	-	-	85.45±0.05	74.68±0.09		
9.5	-	-	90.19±0.09	79.49±0.13		
10	-	-	94.5±0.12	83.19±0.11		
10.5	-	-	98.4±0.14	87.38±0.07		
11	-	-	-	91.35±0.06		
11.5	-	-	-	95.16±0.09		
12	-	-	-	99.45±0.10		



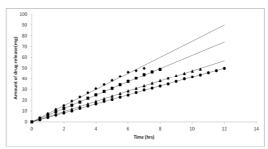


Fig.10.Comparative *in-vitro* drug release profile of Glipizide buccal tablets prepared with different concentrations of Aegle Marmelos gum

Fig.11.Comparative Zero order plots of Glipizide buccal tablets prepared with different concentrations of Aegle Marmelos gum

- ♦ F₁ . Glipizide buccal tablets prepared with Aegle Marmelos gum in 1:05 ratio
- F₂ . Glipizide buccal tablets prepared with Aegle Marmelos gum in 1:0.75 ratio
- ▲ F₃. Glipizide buccal tablets prepared with Aegle Marmelos gum in 1:1 ratio
- ●F₄. Glipizide buccal tablets prepared with Aegle Marmelos gum in 1:1.25

Table.16. *In vitro* drug release kinetic data of Glipizide buccal tablets prepared with different concentrations of Aegle Marmelos gum

Formulation	Zero order	First order	Higuchi	Peppas	T ₅₀ (hr)	T ₉₀ (hr)
F1	0.992	0.954	0.994	0.940	3.0	5.4
F2	0.994	0.967	0.997	0.961	3.9	6.9
F3	0.997	0.974	0.993	0.972	4.9	8.8
F4	0.995	0.989	0.998	0.986	6.2	11.2

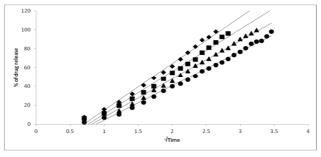


Fig.12.Comparative Higuchi plots of Glipizide buccal tablets prepared with different concentrations of Aegle Marmelos gum

- ◆ F₁ . Glipizide buccal tablets prepared with Aegle Marmelos gum in 1:05 ratio
- F₂ . Glipizide buccal tablets prepared with Aegle Marmelos gum in 1:0.75 ratio
- ▲ F₃. Glipizide buccal tablets prepared with Aegle Marmelos gum in 1:1 ratio
- F₄. Glipizide buccal tablets prepared with Aegle Marmelos gum in 1:1.25 ratio

Mucoadhesive buccal tablets of Glipizide with cashew nut tree gum were prepared by using different drug: gum ratios. The results of the physical characterization of tablets are summarized in Table 17. All the formulations hardness, weight variation, friability and drug content values were found to be within pharmacopoeia limits. The swelling behavior is important for bioadhesion. Water sorption increases with an increase in the concentration of hydrophilic polymers. Swelling index, Mucoadhesive strength and *Ex-vivo* residence time were shown in table 18. The cashew nut tree gum swells slowly and dissolves in the

presence of water. As hydrophilicity of the hydrogel increases, the interaction between water and hydrogel will increase too; this facilitates water diffusion and leads to greater swelling. The surface pH was determined in order to investigate the possibility of any side effects, in the oral cavity as acidic or alkaline pH was bound to cause irritation to the buccal mucosa. Surface pH of all formulations was found to be in the range of 6.45 to 6.72 which were nearer to the salivary pH 6.8 Hence it was assumed that these formulations do not cause any irritation to the mucous layer of the oral cavity.

Table.17.Physical properties of Glipizide buccal tablets prepared with different concentrations of cashew nut

		** ** S *****				
Formulation	Parameters Weight anniation (ma) Handard (halom²) Frieblita (ff) Done contact (ff)					
	Weight variation (mg)	Hardness (kg/cm ²)	Friability (%)	Drug content (%)		
F ₅	250 ± 1	4.3 ± 0.02	0.31	99.56		
F_6	250 ± 3	4.0 ± 0.01	0.48	99.34		
F_7	250 ± 2	4.2 ± 0.03	0.54	99.47		
F_8	250 ± 1	4.1 ± 0.01	0.67	100.02		

Table .18.Mucoadhesion strength, swelling index, retention time, and surface pH of buccal tablets prepared with different concentrations of cashew nut tree gum

united that concentrations of turners had the gain					
Formulation	Swelling index	Ex-vivo mucoadhesion time	Ex-vivo bioadhesivestrength	Surface pH	
F ₅	7.62 ± 3.82	3 hours 10 minutes	15.52 ± 0.32	6.27 ± 0.36	
F_6	8.56 ± 3.60	4 hours 46 minutes	15.86 ± 0.10	6.39 ± 0.07	
F_7	9.61 ± 2.92	6 hours 12 minutes	16.20 ± 0.44	6.48 ± 0.09	
F.	9.95 ± 2.36	9 hours 35 minutes	17.61 ± 1.20	6.79 ± 0.12	

Mucoadhesion is determined by mucoadhesive strength and duration of mucoadhesion. Formulation F₅-F₈ shows good mucoadhesive strength. As the viscosity gum increases swelling increases and mucoadhesion force depends on the swelling of the gum. This improves the consolidation step that increases the mobility of molecule and facilitates the interpretation with mucus layer, thus mucoadhesion increases. F₈ shows maximum mucoadhesive strength this is due to tremendous increase in viscosity. The ex-vivo residence time was determined by using USP disintegration apparatus. Among the four formulations subjected for this study F₈ showed maximum residence time of 9.35 Hrs. It was found that an increase in concentration of the polymer increases the residence time. This was mainly due to the strong mucoadhesion nature of the polymer used. The results of in vitro drug release

studies of different formulation were shown in table 19 and Figure 13. Tablet formulations prepared by using drug and gum in ratios of 1:0.5, 1:0.75, 1:1, and 1:1.25 shown drug release for a period of 6.5 hours, 8.5 hours, 9.5 hours and 11 hours respectively. The initial burst release decrease with increase in concentration of gum. To ascertain the mechanism of drug release, the dissolution data were analyzed by zero order, first order, Higuchi and Peppas equations. The correlation coefficient values (r) and dissolution kinetics values were shown in table 20. Amount of drug release versus time curves exhibited straight line for the formulations and confirmed that the release rate followed zero order release kinetics (Figure 14) percentage of drug release versus the square root of time curves shows linearity and proves that all the formulations followed Higuchi mechanism (Figure 15).

Table 19.In vitro release data of Glipizide buccal tablets Prepared with different concentrations of cashew nut tree gum

Time (hrs)	$F_{5(\%DrugRelease)}$	$F_{6(\%DrugRelease)}$	F _{7 (%Drug Release)}	$F_{8(\%DrugRelease)}$
0	0	0	0	0
0.5	7.34 ± 0.07	5.67±0.05	5.15±0.09	4.48±0.12
1	15.05±0.10	11.58±0.08	10.41±0.13	8.99±0.15
1.5	23.06±0.15	17.51±0.10	15.68±0.11	13.51±0.11
2	30.24±0.11	22.83±0.13	20.98±0.08	18.06±0.9
2.5	38.32±0.09	28.54±0.10	26.07±0.15	22.45±0.16
3	46.01±0.11	34.99±0.14	31.48±0.13	27.04±0.13
3.5	53.18±0.06	41.05±0.17	36.59±0.16	31.64±0.10
4	61.15±0.14	46.71±0.08	41.98±0.12	36.17±0.07
4.5	69.06±0.10	51.50±0.13	47.19±0.07	40.76±0.14
5	76.36±0.08	57.49±0.15	52.21±0.09	45.04±0.05
5.5	84.18±0.14	64.44±0.07	57.51±0.14	49.59±0.12
6	92.26±0.05	69.14±0.09	62.96±0.11	54.86±0.16
6.5	99.34±0.09	75.08±0.11	68.03±0.15	59.04±0.10
7		81.13±0.14	73.24±0.10	63.14±0.08
7.5		87.51±0.16	78.28±0.08	67.96±0.10

8	93.83±0.07	84.04±0.10	72.48±0.07	
8.5	98.18±0.12	89.39±0.05	76.14±0.14	
9		94.56±0.09	81.65±0.16	
9.5		99.67±0.07	86.74±0.11	
10			90.73±0.17	
10.5			95.28±0.09	
11			99.76±0.05	

Table.20: In vitro drug release kinetic data of Glipizide buccal tablets prepared with different concentrations of cashew nut tree gum

Formulation	Correlation coefficient				- T ₅₀ (hr)	T ₉₀ (hr)
roimulation	Zero order	First order	Higuchi	Peppas	- 1 ₅₀ (III)	190 (111)
F_5	0.9969	0.8604	0.9840	0.8408	3.1	5.5
F_6	0.9917	0.8664	0.9865	0.8886	4.3	7.8
F_7	0.9934	0.8673	0.9836	0.9054	4.8	8.6
F_8	0.9962	0.8761	0.9819	0.9139	5.5	9.7

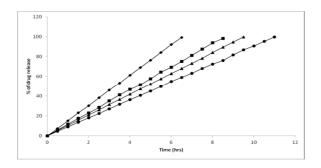


Fig.13. Comparative in vitro drug release profile of Glipizide buccal tablets prepared with different concentrations of cashew nut tree gum

Fig.14. Comparative Zero order plots of Glipizide buccal Tablets prepared with different concentrations of cashew nut tree gum

- \bullet F_5 . Glipizide buccal tablets prepared with Cashew nut tree gum in 1:05 ratio
- F₆. Glipizide buccal tablets prepared with Cashew nut tree gum in 1:0.75 ratio
- ▲ F₇. Glipizide buccal tablets prepared with Cashew nut tree gum in 1:1 ratio
- F₈. Glipizide buccal tablets prepared with Cashew nut tree gum in 1:1.25 ratio

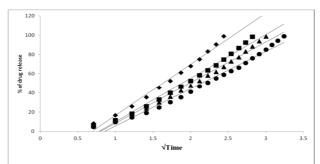


Fig.15. Comparative Higuchi plots of Glipizide buccaltablets prepared with different concentrations of cashew nut tree gum

- ♦ F₅. Glipizide buccal tablets prepared with Cashew nut tree gum in 1:05 ratio
- F₆. Glipizide buccal tablets prepared with Cashew nut tree gum in 1:0.75 ratio
- ▲ F₇. Glipizide buccal tablets prepared with Cashew nut tree gum in 1:1 ratio
 F₈. Glipizide buccal tablets prepared with Cashew nut tree gum in 1:1.25 ratio

Mucoadhesive buccal tablets of Glipizide with Moringa oleifera gum were prepared by using different drug: gum ratios. The results of the physical characterization of tablets are summarized in Table 21. All the formulations hardness, weight variation, friability and drug content values were found to be within pharmacopoeia limits. The swelling behavior is important for bioadhesion. Water sorption increases with an increase in the concentration of hydrophilic polymers. Swelling index, Mucoadhesive strength and Ex-vivo residence time were shown in table 22. The Moringa oleifera gum swells slowly and dissolves in the presence of water. As hydrophilicity of the hydrogel increases, the interaction between water and hydrogel will increase too; this facilitates water diffusion and leads to greater swelling. The surface pH was determined in order to investigate the possibility of any side effects, in the oral cavity as acidic or alkaline pH was bound to cause irritation to the buccal mucosa. Surface pH of all formulations was found to be in the range of 6.12 to 6.62 which were nearer to the salivary pH 6.8 Hence it was assumed that these formulations do not cause any irritation to the mucous layer of the oral cavity. Mucoadhesion is determined by Mucoadhesive strength and duration of mucoadhesion. Formulation F₉-F₁₂ shows good mucoadhesive strength. As the viscosity gum increases swelling increases and mucoadhesion force depends on the swelling of the gum. This improves the consolidation step that increases the mobility of molecule and facilitates the interpretation with mucus layer, thus mucoadhesion increases. F₁₂ shows maximum mucoadhesive strength this is due to tremendous increase in viscosity.

Table.21. Physical properties of Glipizide buccal tablets formulated with different concentrations of Moringa oleifera gum

olehera gum					
Formulation	Parameters				
Formulation	Weight variation (mg)	Hardness (kg/cm²)	Friability(%)	Drug content (%)	
F ₉	250 ± 2	4.2 ± 0.03	0.52	99.38	
F_{10}	250 ± 1	4.1 ± 0.01	0.69	100.05	
F_{11}	250 ± 3	4.4 ± 0.02	0.72	99.45	
F_{12}	250 ± 2	4.5 ± 0.01	0.81	99.16	

Table.22. Mucoadhesion strength, swelling index, retention time, and surface pH of buccal tablets prepared with different concentrations of Moringa oleifera gum

Formulation	Swelling index	Ex-vivo mucoadhesion time	Ex-vivo bioadhesive strength	Surface pH
F ₉	6.86 ± 4.02	3 hours 14 minutes	15.21 ± 0.45	6.12 ± 0.15
F_{10}	7.29 ± 3.90	5 hours 56 minutes	15.75 ± 0.51	6.30 ± 0.10
F_{11}	7.82 ± 3.05	6 hours 45 minutes	16.34 ± 0.36	6.57 ± 0.12
F_{12}	8.30 ± 3.26	8 hours 28 minutes	16.98 ± 0.12	6.62 ± 0.05

The *ex-vivo* residence time was determined using USP disintegration apparatus. Among the four formulations subjected for this study F₁₂ showed maximum residence time of 8.28 Hrs. It was found that an increase in concentration of the polymer increases the residence time. This was mainly due to the strong mucoadhesion nature of the polymer used. The results of *in vitro* drug release studies of different formulation were shown in Table 23 and Figure 16. Tablet formulations prepared by using drug and gum in ratios of 1:0.5, 1:0.75 1:1, and 1:1.25 shown drug release for a period of 6 hours, 8 hours, 9 hours and 10.5 hours respectively. The initial burst release

decrease with increase in concentration of gum. To ascertain the mechanism of drug release, the dissolution data were analyzed by zero order, first order, and Higuchi and Peppas equations. The correlation coefficient values (r) and dissolution kinetics values were shown in table 24. Amount of drug release versus time curves exhibited straight line for the formulations and confirmed that the release rate followed zero order release kinetics (figure 17) percentage of drug release versus the square root of time curves shows linearity and proves that all the formulations followed Higuchi mechanism (figure 18)

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Table.23. *In vitro* release data of Glipizide buccal tablets Prepared with different concentrations of Moringa oleifera gum

olenera gum							
Time (hrs)	$F_{9(\%Drug\;Release)}$	$F_{10(\%Drug\;Release)}$	$F_{11(\%Drug\;Release)}$	$F_{12(\%Drug\ Release)}$			
0	0	0	0	0			
0.5	8.23±0.08	6.13±0.05	5.32±0.09	4.69±0.11			
1	16.59±0.12	12.45±0.09	11.03±0.11	9.45±0.15			
1.5	24.79±0.06	18.23±0.14	16.31±0.07	14.41±0.17			
2	33.26±0.13	24.91±0.06	21.78±0.16	18.92±0.08			
2.5	41.37±0.09	30.87±0.13	26.91±0.11	23.65±0.10			
3	49.54±0.17	35.89±0.08	32.86±0.08	28.21±0.15			
3.5	58.03±0.14	43.42±0.15	37.52±0.05	33.29±0.07			
4	64.54±0.11	49.56±0.09	43.63±0.16	37.96±0.012			
4.5	74.99±0.06	55.29±0.11	49.54±0.11	42.67±0.05			
5	83.03±0.08	62.46±0.16	54.13±0.08	47.42±0.13			
5.5	91.37±0.05	67.86±0.07	60.82±0.13	52.16±0.08			
6	99.78±0.09	74.67±0.15	66.20±0.07	57.06±0.10			
6.5		80.64±0.06	71.86±0.12	61.71±0.15			
7		86.57±0.11	77.11±0.15	66.41±0.11			
7.5		93.46±0.09	83.09±0.11	71.15±0.17			
8		99.33±0.015	88.21±0.09	76.10±0.13			
8.5			93.83±0.14	80.52±0.07			
9			98.82±0.07	85.15±0.05			
9.5				89.28±0.12			
10				95.13±0.06			
10.5				99.73±0.14			

Table.24. In vitro drug release kinetic data of Glipizide Buccal tablets prepared with Moringa oleifera gum

Formulation	Correlation coefficient				T (hr)	T (hm)
	Zero order	First order	Higuchi	Peppas	$-T_{50}\left(hr\right)$	T ₉₀ (hr)
F ₉	0.9945	0.8603	0.9887	0.8203	3.0	5.4
F_{10}	0.9929	0.8790	0.9818	0.8790	4.0	7.2
F_{11}	0.9919	0.8971	0.9834	0.8986	4.5	8.2
F_{12}	0.9962	0.9107	0.9855	0.9107	5.3	9.2

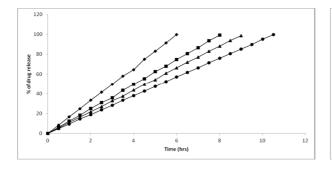


Fig.16. Comparative *in vitro* drug release profile of Gipizide buccal tablets prepared with different concentrations of Moringa oleifera gum

Fig.17. Comparative Zero order plots of Glipizide buccal tablets prepared with different concentrations of Moringa oleifera gum

- ♦ F₉. Glipizide buccal tablets prepared with Moringa oleifera gum in 1:05 ratio
- F₁₀. Glipizide buccal tablets prepared with Moringa oleifera gum in 1:0.75 ratio
- ▲ F₁₁. Glipizide buccal tablets prepared with Moringa oleifera gum in 1:1 ratio
- F₁₂-Glipizide buccal tablets prepared with Moringa oleifera gum in 1:1.25 ratio

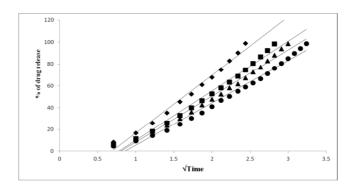


Fig.18.Comparative Higuchi plots of Glipizide buccal tablets prepared with different concentrations of Moringa oleifera gum

- ♦ F₉. Glipizide buccal tablets prepared with Moringa oleifera gum in 1:05 ratio
- F₁₀. Glipizide buccal tablets prepared with Moringa oleifera gum in 1:0.75 ratio
- ▲ F₁₁. Glipizide buccal tablets prepared with Moringa oleifera gum in 1:1 ratio
- F₁₂-Glipizide buccal tablets prepared with Moringa oleifera gum in 1:1.25 ratio

Conclusion

- 1. Glipizide buccal tablets prepared with natural polymers such as *Aegle marmelos gum*, *Cashew nut tree* gum and *Moringa oleifera* gum has shown the prolonged release.
- 2. Among the three polymers, Aegle marmelos shows more prolonged release compared with other polymers (Aegle marmelos > Cashew nut tree gum > Moringa oleifera gum).
- 3. Glipizide buccal tablets prepared with aegle marmelos gum in 1:3 ratios shows more prolonged drug release compared with the other polymers (1:3 > 1:2 > 1:1).
- 4. The prepared Glipizide buccal tablets compiles with the Indian Pharmacopeia standards.
- 5. Surface pH of all formulations was found to be in the range of 6.12 6.72, which were nearer to the salivary pH 6.8. Hence it was assumed that these formulations do not cause any irritation to the mucous layer of the oral cavity.
- 6. It was found that an increase in concentration of the polymer increases the *ex vivo* Mucoadhesive residence time.
- 7. As the viscosity gum increases swelling increases and mucoadhesion force depends on the swelling of the gum.
- 8. FTIR and DSC studies clearly indicate that there is no drug polymer interaction.
- 9. All the formulations drug release followed zero order kinetics and the mechanism of the drug release was governed by Higuchi model.
- By consideration of all above parameters, it that *Aegle marmelos* gum appears to be suitable for use as a release retardant in the manufacture of buccal tablets because of its good swelling, good flow rate and suitability for mucoadhesion formulations. From the

dissolution study, it was concluded that dried Aegle marmelos gum can be used as an excipient for preparing Mucoadhesive buccal tablets.

"Cite this Article"

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References

1.Alur H.H, Johnston T.P, Mitra A.K. Peptides and Proteins: Buccal Mucosa. In. J. Superbrick, J.C. Boylan (Eds.), Encyclopedia of Pharmaceutical Technology. 1st edition, Marcel Dekker Inc, New York, 2001; 20 (3): 193–218.

2.Rathbone M.J, Ponchel G, Ghazali F.A. Systemic and oral mucosal drug delivery and delivery systems. In: M.J. Rathbone (Ed.), Oral Mucosal Drug Delivery, Marcel Dekker Inc New York, 1996; 74: 241–284.

3.Harris D, Robinson J.R. Drug delivery via the mucous membranes of the oral cavity. J. Pharm. Sci 1992; 81:1–10.

4.Gandhi R.B, Robinson J.R. Bioadhesion in drug delivery. Ind. J. Pharm. Sci 1988; 50 (3): 145–152.

5.Squier C.A, Finkelstein M.W, in: A.R. Ten Cate (Ed.). Oral Histology, Development, Structure and Function, C.V. Mosby, St. Louis 1989; 345–385.

6.Hao J, Heng P.W.S. Buccal delivery systems. Drug Dev. Ind. Pharm 2003; 29 (8): 821–832.

7.Shojaei A.H, Ling X. Determination of transport route of acyclovir across buccal mucosa. Proc. Int.

- Symp. Control. Release Bioact. Mater 1997; 24: 427–428.
- 8.Chen. L, Hui-Nan X, Xiao-Ling L. In vitro permeation of tetramethylpyrazine across porcine buccal mucosa. Acta Pharmacol. Sin 2002; 23: 792–796.
- 9.Nielsen H.M, Rassing M.R. TR146 cells grown on filters as a model of human buccal epithelium. III. Permeability enhancement by different pHvalue, different osmolarity value, and bile salts. Int. J. Pharm 1999; 185: 215–225.
- 10.Zhang H, Robinson J.R. In vitro methods for measuring permeability of the oral mucosa. In: J. Swarbrick, J.C. Boylan (Eds.), Oral Mucosal Drug Delivery, 1st edition, Marcel Dekker, INC, New York 1996; 74: 85–100.
- 11.Mashru R, Sutariya V, Sankalia M. J. Sankalia. Transbuccal delivery of lamotrigine across porcine buccal mucosa: in vitro determination of routes of buccal transport. J.Pharm. Pharmaceut. Sci 2005; 8 (1): 54–62.
- 12.Randhawa M.A, Malik S.A, Javed M. Buccal absorption of weak acidic drugs is not related to their degree of ionization as estimated from the Henderson–Hasselbalch equation. Pak. J. Med. Res 2003; 42 (2): 60 65.
- 13.Utoguchi N, Watanabe Y, Suzuki T, Maeharai J, Matsumoto Y. Pharm. Res 1997; 14: 320–324.
- 14.Nielsen H.M, Rassing M.R. TR146 cells grown on filters as a model of human buccal epithelium. III. Permeability enhancement by different pH value, different osmolarity value, and bile salts, Int. J. Pharm 1999; 185: 215–225.
- 15.Odeblad E. The discovery of different types of cervical mucus, Bull. Ovul. Method Res. Ref. Cent. Aust 1994; 21: 3–35.
- 16.Lowman A, Peppas N.A. Complexation graft copolymers as oral drug delivery systems. Polym. Prepr 1997; 38 (2): 566–567.
- 17. Weathercell J.A, Robinson C, Rathbone M.J. Site-specific differences in the salivary concentrations of substances in the oral cavity implications for the etiology of oral-disease and local drug delivery. Adv. Drug Del. Rev 1994; 130: 24-42.
- 18. Veuillez F, Kalia Y.N, Jacques Y, Deshusses J, Buri P. Factors and strategies for improving buccal absorption of peptides. Eur. J. Pharm. Biopharm 2001; 51 (2): 93–109.
- 19.Longer M.A, Robinson J.R. Fundamental aspects of bioadhesion. Pharm. Int 1986; 7: 114–117.
- 20.Patil D.N, Kulkarni AR, Hatapakki B.C and Patil BS. Preparation and Evaluation of Aegle marmelos Gum as Tablet Binder. International Journal of Pharma and Bio Sciences 2010; 1(1):15.

- 21.Gowthamarajan. K, phani kumar K, Narayan Babulal G, Suresh. B. Preliminary study of Anacardium occidentale gum as binder in formulation of paracetamol tablets. Carbohydrate polymer 2011; 83: 506 511.
- 22.Baswaraj S, Srinivas R, Upendra k, korwar Prakash G. Formulation of Moringa Oleifera gum as a binder in tablet formulation. International journal of Research in Ayurveda and Pharmacy 2010; 1(2): 590-596.
- 23. Michael E Aulton, Pharmaceutics: The science dosage form Design, 2nd Edition, Churchill Livingstone, 2006; 205 206.
- 24. S.A. Mortazavi, J.D. Smart, An investigation of some factors influencing the *in vitro* assessment of mucoadhesion. Int. J. Pharm. 116 (1995) 223–230.
- 25. H.K. Goud, T.M.P. Kumar, Preparation and evaluation of a novel buccal adhesive systems, AAPS Pharm. Sci. Tech. 5(3),(2004),35.
- 26.Martin A, Swarbrick J. Physical pharmacy, 3rd edition, Varghese publishing house, Mumbai, 1993 444 447
- 27. Saikat Pande, Marina Koland, Jolly R Parikh, Ajay B. Solanki, Gaurav Negi, Rahul Trivedi. Buccoadhesive Tablets of Losartan Potassium: Design and Characterization. Int J Pharm and Bio Arc 2010; 2(1): 150-154.
- 28. Venkatchalam Raju P, Goverdhan Reddy P. Formulation and in vitro evaluation of buccal tablets of metoprolol succinate. Int. Res J Pharm. App Sci 2013; 3(2):102-111.
- 29. Vishnu Y, Ramesh G, Chandrasekhar K. In-vitro dissolution profile, in vitro permeation studies. Acta pharm 2007; 2 (3): 185-196.
- 30.Parvez N, Ahuja A, Khar RK. Development and evaluation of mucoadhesive buccal tablets of Lignocaine Hydrochloride, Ind J Pharm Sci 2002; 64(6): 563-567.
- 31. Satyabrata Bhanja, P. Ellaiah, Sujit Kumar Martha, Pratit Kanchan Sahu, Sandip Prasad Tiwari, Bibhuti Bhusan Panigrahi, Debajyoti Das. Formulation and in vitro evaluation of mucoadhesive buccal tablets of Timolol maleate. Int J Pharm Biomed Res 2010; 1(4): 129-134.
- 32.Pandey Sonia, Gupta Arti, Yadav Jitendra Singh, Shah D.R. Formulation and in-vitro evaluation of bilayered buccal tablets of carvediol. Indian J Pharm Edu Res 2010; 44(3): 1-8.
- 33.Nakhat PD, Kondawar AA, Rathi LG, Yeole PG. Development and invitro evaluation of mucoadhesive tablets of metoprolol tartarate. Ind J Pharm Sci.2008; 70(1): 121-125.
- 34. Beckett AH and Stenlake JB: Practical Pharmaceutical Chemistry, Part II. CBS Publishers, Delhi, Fourth Edition 2004: 72-75.

73